

NanoFASE Deliverable D6.3

Report on the laboratory tests on surface transformation and atmospheric chemical reactions

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Research Report Summary

TiO₂ and CeO₂ are among the most produced engineered nanomaterials (ENM). TiO₂ is often used because of its known photocatalytic properties, whereas CeO₂ is used as a diesel fuel additive. TiO₂ can for example photocatalytically reduce NO₂ to NO (see Figure 1). Ambient NO₂ stems mainly from combustion processes in traffic and industry and can be harmful to human health upon inhalation. The aim of the work presented here was to study the effect of TiO₂ and CeO₂ released into the atmosphere on atmospheric NO₂ concentrations under UV exposure. The outcome shall be used in the multimedia model in NanoFASE for the atmospheric compartment. Our results show that the reaction of TiO₂ with NO₂ is a function of the TiO₂ concentrations as well as the UV exposure. At high particle concentrations, the reduction of the NO₂ concentration was nearly independent of the UV exposure. On the other hand, for high UV exposure, the effect of the TiO₂ concentration is very small. Most importantly, TiO₂ only showed a noticeable effect on the NO₂ concentrations at TiO₂ particle concentrations $>5 \times 10^5 \text{ \#/cm}^3$, which are very unlikely in the atmosphere, except for locations near the particle release. The effect of CeO₂ ENM on atmospheric NO₂ concentrations was found to be negligibly small. In summary, it can be expected that the effect of TiO₂ and CeO₂ on ambient NO₂ concentrations can be expected to be generally very small if not negligible due to too low ambient concentrations of TiO₂ and CeO₂ nanomaterials.



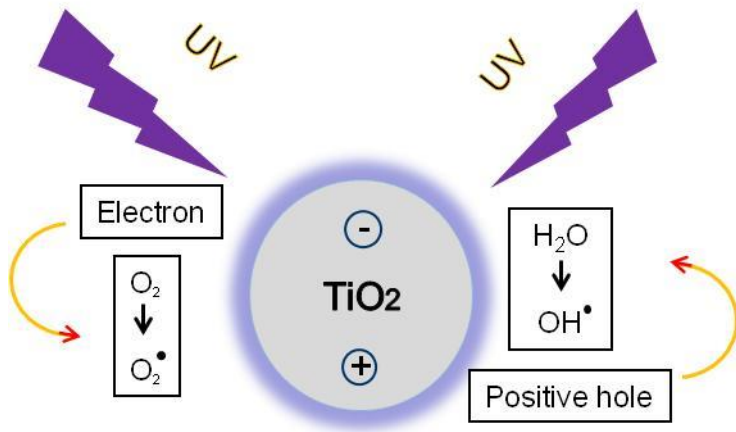


Figure 1: Mechanism of photocatalytic reaction with TiO₂ particles

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