

## NanoFASE Deliverable D6.3

### Report on the laboratory tests on surface transformation and atmospheric chemical reactions

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#### Research Report Summary

TiO<sub>2</sub> and CeO<sub>2</sub> are among the most produced engineered nanomaterials (ENM). TiO<sub>2</sub> is often used because of its known photocatalytic properties, whereas CeO<sub>2</sub> is used as a diesel fuel additive. TiO<sub>2</sub> can for example photocatalytically reduce NO<sub>2</sub> to NO (see Figure 1). Ambient NO<sub>2</sub> stems mainly from combustion processes in traffic and industry and can be harmful to human health upon inhalation. The aim of the work presented here was to study the effect of TiO<sub>2</sub> and CeO<sub>2</sub> released into the atmosphere on atmospheric NO<sub>2</sub> concentrations under UV exposure. The outcome shall be used in the multimedia model in NanoFASE for the atmospheric compartment. Our results show that the reaction of TiO<sub>2</sub> with NO<sub>2</sub> is a function of the TiO<sub>2</sub> concentrations as well as the UV exposure. At high particle concentrations, the reduction of the NO<sub>2</sub> concentration was nearly independent of the UV exposure. On the other hand, for high UV exposure, the effect of the TiO<sub>2</sub> concentration is very small. Most importantly, TiO<sub>2</sub> only showed a noticeable effect on the NO<sub>2</sub> concentrations at TiO<sub>2</sub> particle concentrations  $>5 \times 10^5 \text{ \#/cm}^3$ , which are very unlikely in the atmosphere, except for locations near the particle release. The effect of CeO<sub>2</sub> ENM on atmospheric NO<sub>2</sub> concentrations was found to be negligibly small. In summary, it can be expected that the effect of TiO<sub>2</sub> and CeO<sub>2</sub> on ambient NO<sub>2</sub> concentrations can be expected to be generally very small if not negligible due to too low ambient concentrations of TiO<sub>2</sub> and CeO<sub>2</sub> nanomaterials.



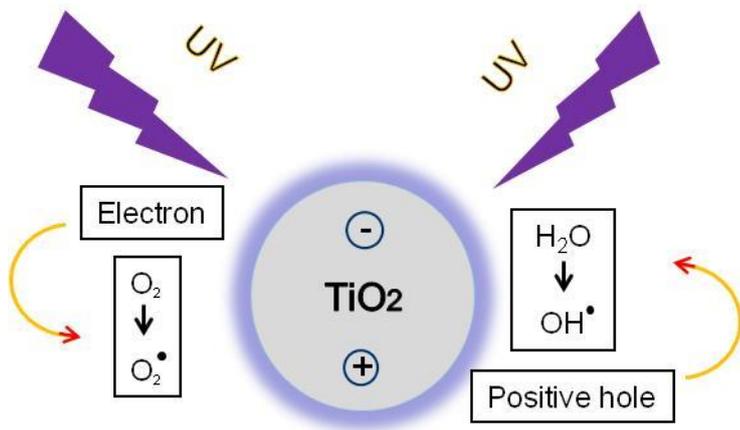


Figure 1: Mechanism of photocatalytic reaction with TiO<sub>2</sub> particles

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Foreseen publication date of the full deliverable: [for CO deliverable; if possible with some explanation why the deliverable is currently confidential (e.g. waiting for scientific publication)]

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March 2019



The NanoFASE project receives funding from the European Union's Horizon 2020 research and innovation programme under grant agreement number 642007. This publication reflects only the authors' view and the Commission is not responsible for any use that may be made of the information it contains.